



Key roles of Arg⁵, Tyr¹⁰ and His residues in A β -heme peroxidase: Relevance to Alzheimer's disease



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ABSTRACT

Recent reports show that heme binds to amyloid β -peptide (A β) in the brain of Alzheimer's disease (AD) patients and forms A β -heme complexes, thus leading a pathological feature of AD. However, the important biological relevance to AD etiology, resulting from human A β -heme peroxidase formation, was not well characterized. In this study, we used wild-type and mutated human A β_{1-16} peptides and investigated their A β -heme peroxidase activities. Our results indicated that both histidine residues (His¹³, His¹⁴) in A β_{1-16} and free histidine enhanced the peroxidase activity of heme, hence His residues were essential in peroxidase activity of A β -heme complexes. Moreover, Arg⁵ was found to be the key residue in making the A β_{1-16} -heme complex as a peroxidase. Under oxidative and nitrative stress conditions, the A β_{1-16} -heme complexes caused oxidation and nitration of the A β Tyr¹⁰ residue through promoting peroxidase-like reactions. Therefore, these residues (Arg⁵, Tyr¹⁰ and His) were pivotal in human A β -heme peroxidase activity. However, three of these residues (Arg⁵, Tyr¹⁰ and His¹³) identified in this study are all absent in rodents, where rodent A β -heme complex lacks peroxidase activity and it does not show AD, implicating the novel significance of these residues as well as human A β -heme peroxidase in the pathology of AD.

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1. Introduction

Alzheimer's disease (AD) is one of the most common progressive neurodegenerative disorders in the elderly. Many studies have demonstrated that excess amyloid β -peptides (A β) in the brain are believed to be the culprits in the neurodegeneration of AD [1–3]. The aggregation of A β peptides and generation of reactive oxygen species (ROS) are the two markers of AD and can be responsible for the early oxidative damage observed in AD [3,4]. The role of metal ions (e.g., Cu²⁺, Fe³⁺, and Zn²⁺) in modulating the A β aggregation and in generating ROS is being actively investigated due to the fact that brains of AD patients contain abnormally higher levels of metal ions [3,5–7].

Heme, which is a ferroporphyrin IX complex, is essential to the function of a number of proteins. Recent studies show that heme binds to A β to form an A β -heme complex, which can stabilize

the structure of A β and inhibit A β aggregation [8–11]. However, depletion of biologically required heme by A β binding can result in symptoms such as increases in heme synthesis and iron uptake, abnormal iron homeostasis, dysfunction in mitochondrial complex IV, and oxidative stress, etc [8]. On the other hand, the A β -heme complex also exhibits increased peroxidase activity with respect to free heme, and can catalyze the oxidation of specific neurotransmitters (such as 3,4-dihydroxyphenylalanine, serotonin) by H₂O₂ [8,9]. This peroxidase activity could be a probable reason for the oxidative damage and abnormal neurotransmission observed in AD patients. The above symptoms are the characteristic pathological features of AD, and thus the formation of A β -heme complex opens up a new dimension in AD pathologic research.

Atamna et al. compared heme-binding between human A β and rodent A β , and found that human A β , unlike rodent (i.e., mouse, rat, etc.) A β , tightly bound to heme and formed a peroxidase-like complex. Although both human A β and rodent A β could form aggregates equally, rodents lack AD-like neuropathology [12]. These findings suggest that formation of A β -heme peroxidase contributes to human A β 's neurotoxicity and the increased human susceptibility to AD. The amino acid sequence of rodent A β is identical to that of human A β except for three amino acids (Arg⁵Gly⁵, Tyr¹⁰Phe¹⁰, His¹³Arg¹³) within the hydrophilic region (Fig. 1A), which

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implies possible important roles of the three residues of human A β in AD pathology [12]. Furthermore, these amino acids Arg, Tyr, and His are found to participate in heme-binding in heme-proteins and peroxidases [12–15], which drive us to propose that A β -heme peroxidase is a key molecular link between these residues present in human A β and the increased human susceptibility to AD. Recent results showed that Arg⁵ residue was required for A β -heme peroxidase activity and His¹³ and His¹⁴ residues were the heme-binding ligands [16,17].

It is now well established that heme binds to A β peptides and the adducts of A β -heme complex show higher peroxidase activity than free heme, but the important biological relevance to AD etiology, resulting from human A β -heme peroxidase formation, have not been well characterized. In this study, we used wild-type and mutated human A β_{1-16} peptides and investigated their A β -heme peroxidase activities, and found that Arg⁵ and His residues (His¹³, His¹⁴) were critical for human A β -heme peroxidase activity. In the presence of oxidative stress (H₂O₂), A β_{1-16} -heme complex produced dimerization of the peptide through dityrosine cross-linking, and in addition A β underwent endogenous nitration at Tyr¹⁰ when nitrite (NaNO₂) was also present. This result was important for the potential pathogenic role of Tyr¹⁰ in AD, because both dityrosine cross-linking and Tyr¹⁰ nitration critically accelerated A β aggregation and plaque formation [5,14,15], thus implying that the formation of A β -heme peroxidase complex under oxidative stress conditions could be a promoting factor for the A β aggregation process. Hence, three of these residues (Arg⁵, Tyr¹⁰ and His¹³) absent in rodent A β and formation of human A β -heme peroxidase can play a vital role in AD pathology, while rodent A β -heme complex lacks peroxidase activity and it does not show AD.

2. Materials and methods

2.1. Materials

L-Histidine (His), L-arginine, ferriprotoporphyrin IX chloride (hemin, which is referred to as “heme” here), and 2,2'-Azinobis (3-ethylbenzothiazoline-6-sulphonic acid) diammonium salt (ABTS) and rabbit polyclonal antibody against 3-nitrotyrosine were purchased from Sigma. Soluble A β peptides (A β_{1-16} , A β_{17-40} , and A β_{1-40}) and the mutated peptides were synthesized by GL Biochem (Shanghai, China) with >95% purity. The A β_{1-16} mutants used were Arg⁵Asn (R5 N), His¹³Ala (H13A), His¹⁴Ala (H14A), and double mutant His¹³Gly, His¹⁴Gly (H13G-14G).

2.2. Binding of A β with heme

A β peptides were dissolved in 16% (v/v) CH₃CN/water, while heme solution was prepared in dimethyl sulfoxide [11]. The stock of heme and A β was diluted in 100 mM phosphate-buffered saline (PBS, pH 7.0). A β -heme complexes were prepared by incubating 1 equiv of both heme and A β solutions at room temperature (~25 °C) for 30 min. Then, the complexes were used for peroxidase activity determination.

2.3. Peroxidase activity measurement

ABTS was used as the substrate to measure the peroxidase activity of heme [11,15]. The assay mixture (in PBS, pH 7.0) contained 5 μ M heme, 1 mM ABTS, 0.5 mM H₂O₂, and in the presence or absence of A β (or free histidine). The peroxidase activity was measured by monitoring the increase in absorbance at 734 nm. The values were the absorptions subtracted by that at 0 min. The kinetic constants (k_{obs}) were obtained from the initial rates of the reactions and calculated from absorbance vs. time [15].

2.4. Dimerization and nitration of A β

In the presence or absence of sodium nitrite (NaNO₂, 1 mM), the reaction was carried out by adding H₂O₂ (1 mM) to a solution containing heme (20 μ M) and A β_{1-16} (100 μ M) in PBS (pH 7.0). The mixture was incubated for 60 min at 37 °C in the dark and then analyzed for dimerization and nitration of A β_{1-16} . Significant protein modifications were observed in short time incubation when high concentrations of heme-H₂O₂-NO₂⁻ were used in many in vitro experiments [10,11,15]. These high concentrations were, therefore, chosen in our studies to conveniently compare the different effects of A β -heme complexes.

Dityrosine (3,3'-dityrosine), an oxidation product of tyrosine produced by reaction between tyrosyl radicals, is a highly stable marker of tyrosyl radical activity and an intensely fluorescent compound. The formation of dityrosine (A β -A β dimeric peptide) was analyzed by measuring fluorescence spectra [5]. Dityrosine fluorescence was excited at 320 nm and monitored at 350–500 nm. Fluorescence intensity was measured at 410 nm. The spectra were recorded with a fluorescence spectrophotometer PerkinElmer-LS55.

Dot blotting was usually performed in detecting protein tyrosine nitration [14]. A rabbit polyclonal antibody against 3-nitrotyrosine was used for detection of the nitrated A β peptide in this study.

2.5. Statistical analysis

All of the experiments were performed at least three times. The results were reported as the means \pm SD of at least triplicate determinations. One-way ANOVA was used for statistical analyses, and $p < 0.05$ was considered significant.

3. Results and discussion

3.1. His residues were essential in peroxidase activity of A β -heme complexes

It is well-known that heme binds to A β peptides in the hydrophilic portion (residues 1–16) and His residues are found to be the possible binding ligands [16–18]. Thus, A β_{1-40} , A β_{1-16} and A β_{17-40} fragments were used to investigate the possible relationship between histidine-bound complex and peroxidase-like active site. By following the catalytic oxidation of the substrate ABTS by H₂O₂, the peroxidase activities of free heme and A β -heme complexes were investigated. Compared with the free heme, both A β_{1-40} -heme and A β_{1-16} -heme complexes showed obviously higher and similar peroxidase activities, while A β_{17-40} -heme complex showed no enhanced peroxidase activity (Fig. 1B). The result further indicated that the hydrophilic N-terminal of A β was involved in heme binding [16–18], and played an important role in promoting the peroxidase activity of heme. This peroxidase activity of A β -heme complexes (with k_{obs} of $4.2 \times 10^{-3} \text{ min}^{-1}$) was ~3 times faster compared to that of free heme (with k_{obs} of $1.5 \times 10^{-3} \text{ min}^{-1}$), similar to a previous study [16].

Recent studies have shown that His¹³ and His¹⁴ residues in the N-terminal hydrophilic region of A β are the heme-binding ligands, and His¹³ binds to the iron center of heme preferentially when both residues are present [16–18]. Due to the similar active site environments of the heme-bound complexes in A β_{1-40} and A β_{1-16} , A β_{1-16} and site-directed mutants of A β_{1-16} including single mutants (His¹³Ala, His¹⁴Ala), and double mutant (His¹³Gly, His¹⁴Gly) were thus used to observe the effect of His residues on peroxidase activity of A β -heme complexes. As shown in Fig. 1C, the double mutant (His¹³Gly, His¹⁴Gly) showed no enhanced

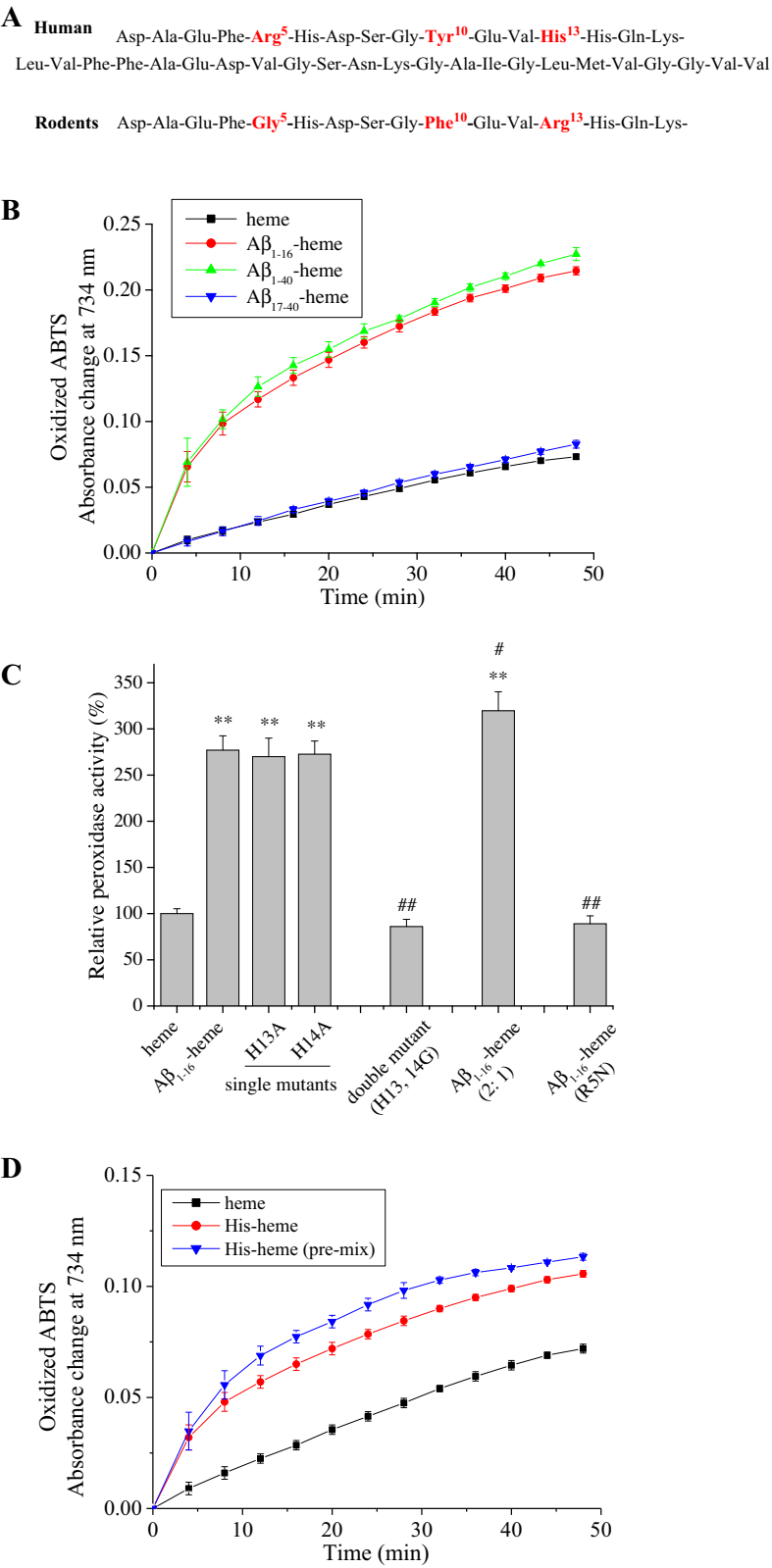


Fig. 1. (A) Amino acid sequence of human A β_{1-40} and rodent A β_{1-16} . (B) Effects of different A β peptides on the peroxidase activity of heme. The peroxidase activities were evaluated by relative ABTS oxidation, monitoring the increase of the 734 nm absorbance intensity. ABTS (1 mM) was treated with heme (5 μ M)-H $_2$ O $_2$ (0.5 mM) (control), and in the presence of different A β peptides (A β_{1-16} , A β_{17-40} , A β_{1-40}) binding. (C) Effect of His and Arg residues in A β_{1-16} on heme peroxidase activity. The reaction mixtures (in 0.1 M PBS) containing ABTS (1 mM), H $_2$ O $_2$ (0.5 mM), heme (5 μ M, control group), and in the presence of different A β_{1-16} peptides (single, double His mutants, and Arg⁵ mutant, 5 μ M, A β_{1-16} :heme = 1:1) binding with different concentrations (A β_{1-16} :heme = 1:1 or 2:1) were incubated at 37 $^{\circ}$ C for 2 h. The corresponding analysis for relative peroxidative activity was obtained. (D) Effects of free His on the peroxidase activity of heme. ABTS (1 mM) was treated with heme (5 μ M)-H $_2$ O $_2$ (0.5 mM) (control), and in the presence or absence of His. His-heme group represented the control group plus free His (50 μ M), His-heme (premixed) group represented the control group plus pre-reacted His-heme (50 μ M) (t = 10 min). The values were the absorption subtracted by that at 0 min and presented as means \pm SD of three independent experiments. The respective control values were set to 100%, to which the other values were compared. ** p < 0.01 compared to control group, ## p < 0.01, * p < 0.05 compared to A β_{1-16} -heme group.

peroxidase activity compared to free heme. The result demonstrated that histidine residues (His¹³, His¹⁴) in A β _{1–16} were essential in peroxidase activity of A β _{1–16}-heme complex and the formation of histidine-bound A β complex had high peroxidase-like active site, which was consistent with the fact that A β _{1–16} with the double mutant did not bind heme [16,17]. The single mutants of A β _{1–16} (His¹³Ala and His¹⁴Ala) showed peroxidase activity comparable to that of the wild-type A β _{1–16} peptide (Fig. 1C). This implied that the changing of the coordinating histidine residues did not significantly affect the peroxidase activity of A β -heme complex, although the active sites were not identical for His¹³- and His¹⁴-bound heme complexes. In addition, the peroxidase activity of A β _{1–16}-heme complexes increased slightly with an increase in the ratio of A β _{1–16} to heme (from 1:1 to 2:1). This data suggested that the coordinated histidine residues in A β _{1–16}-heme complexes did not inhibit the interaction between the iron(III) center and H₂O₂, possibly due to the lability of the iron axial ligands [15].

Furthermore, free His was selected to investigate the important role of His on heme peroxidative activity. Similar to the peroxidase activity of A β -heme complexes (Fig. 1C), the presence of free His enhanced catalytic activity of heme (Fig. 1D). Moreover, the premix of His and heme for 10 min before adding to the reaction mixture showed more significant enhancement on heme peroxidase activity than when adding His and heme to the reaction mixture one by one (Fig. 1D), further demonstrating that the binding of free His could promote heme catalytic activity. The similar enhancement on heme peroxidase activity was observed for the addition of free imidazole (data not shown). Taken together, these results herein indicated that both histidine residues (His¹³, His¹⁴) in A β _{1–16} and free histidine were essential for the higher catalytic activity of heme, implying that the A β peptide bound to heme through histidine ligands and subsequently enhanced the peroxidase activity of heme.

3.2. Arginine residue was responsible for the peroxidase activity of A β -heme complexes

It is generally accepted that an acidic arginine residue exists in the distal side of heme and can serve as the proton source required for peroxidase activity [13,16,17]. As expected, A β _{1–16}(Arg⁵Asn)-heme complex showed no enhanced activity compared to free heme (Fig. 1C). This clearly demonstrated that Arg⁵ was important

in making A β _{1–16}-heme complex as a peroxidase. However, no increase in peroxidase activity occurred upon adding free arginine to heme, His-heme complex or A β _{1–16}-heme complex (Fig. 2). This result indicated that the catalytic role of Arg in peroxidase activity was merely present in the A β -heme complex and related to the intact structure of A β -heme. In catalytic cycle of A β -heme complex, the Arg⁵ residue present in A β could act as an acid catalyst to promote the cleavage of the peroxide O–O bond, hence making the A β -heme complex function as a peroxidase [16,17]. It is important to note that rodent A β , which lacks this Arg⁵ residue (Fig. 1A), shows no significant enhancement in peroxidase activity relative to free heme [12], similar to A β _{1–16}(Arg⁵Asn) peptides (Fig. 1C). The result was also consistent with the fact that Arg⁵ was the key distal residue required for the A β -heme complex to function as a peroxidase.

3.3. Dimerization and nitration of A β _{1–16} under oxidative and nitrative stress

Given that the peroxidase activity of A β -heme complexes toward external substrates was widely observed, we then investigated whether this peroxidase reactivity could be important when addressed on the endogenous peptide. The characteristic fluorescence spectra of dihydroxytyrosine at 410 nm were assessed for tyrosine oxidative modification in A β [5]. Upon reaction of A β _{1–16}-heme with H₂O₂, analysis of the peptide showed an increase in the characteristic fluorescence for tyrosine cross-links (Fig. 3A). The dihydroxytyrosine-contained peptide was only observed in the presence of both

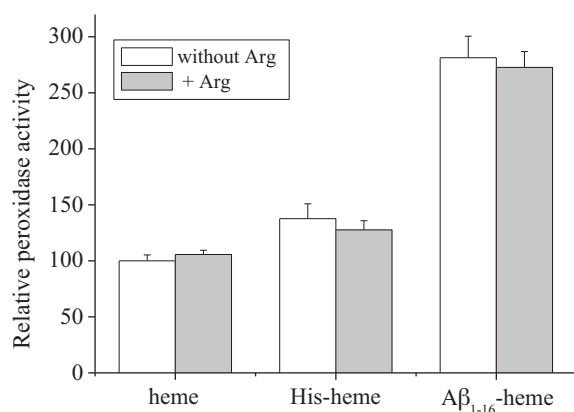


Fig. 2. Potential effect of free arginine on A β -heme peroxidase activity. The reaction mixtures (in 0.1 M PBS) containing ABTS (1 mM), H₂O₂ (0.5 mM), heme (5 μ M, control group, or His-heme (50 μ M), A β _{1–16}-heme (5 μ M)), and in the presence or absence of free L-arginine (50 μ M) were incubated at 37 °C for 2 h. The values were the absorption subtracted by that at 0 min and presented as means \pm SD of three independent experiments. The respective control values were set to 100%, to which the other values were compared.

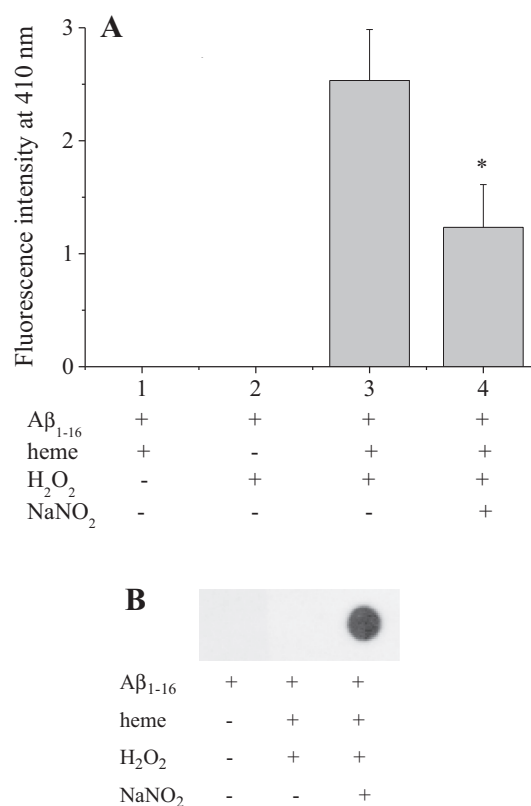


Fig. 3. Dimerization and nitration of A β _{1–16} under oxidative and nitrative stress. The mixture containing heme (20 μ M), A β _{1–16} (100 μ M), H₂O₂ (1 mM), and in the presence or absence of NaNO₂ (1 mM), was incubated in PBS (pH 7.0) for 60 min. (A) The fluorescence intensity of dihydroxytyrosine formation was measured at 410 nm. The values presented as means \pm SD of three independent experiments. **p* < 0.05 compared to control group (heme-H₂O₂ treated, group 3). (B) Tyrosine nitration was determined by dot blotting using 3-nitrotyrosine antibody.

heme and H_2O_2 . The formation of dityrosine was also confirmed by the dot blot of the peptide by using dityrosine antibody (data not shown). The presence of Tyr¹⁰ in A β enabled us to infer the formation of 3,3'-dityrosine cross-link between the two A β_{1-16} chains. In heme peroxidase cycle, ferryl intermediates can oxidize tyrosine residues in A β to highly reactive tyrosyl radical, which can rapidly react with another tyrosyl radical and recombine to dityrosine-contained dimeric peptide (Fig. 4). The similar types of cross-link were previously observed by peroxynitrite-induced dityrosine cross-linked A β [14], and the copper(II)- H_2O_2 -mediated oxidation of A β [5]. A β -A β cross-linking via the tyrosine residues increased the structural strength of A β , and subsequently made the peptide highly resistant to proteolysis and promoted its oligomerization [5,15].

The reactivity of A β_{1-16} -heme complex towards H_2O_2 was further investigated in the presence of NaNO_2 , which typically simulated the oxidative and nitrative stress. Fluorescence spectra analysis at 410 nm showed a decrease in the amount of dityrosine-contained peptide (Fig. 3A). Moreover, heme- H_2O_2 - NaNO_2 system was a classic tyrosine nitrating agent [19,20], indicating that A β would undergo tyrosine nitration. This was confirmed by the dot blot of the peptide by using 3-nitrotyrosine antibody, and showed the formation of nitrated tyrosine (Fig. 3B). In the absence of NaNO_2 , Tyr¹⁰ nitration was not observed. In heme peroxidase pathways, the ferryl intermediates can oxidize NaNO_2 and tyrosine residues in A β to the concomitant formation of nitrogen dioxide ($\cdot\text{NO}_2$) and tyrosyl radicals respectively, which combine to form 3-nitrotyrosine (Fig. 4) [21]. The formation of 3-nitrotyrosine competes with the dimerization of tyrosyl radicals

to 3,3'-dityrosine. Thus, the presence of NaNO_2 showed an inhibitive effect in A β -heme- H_2O_2 -induced dityrosine-contained peptide formation (Fig. 3A). A similar tyrosine nitration was observed by peroxynitrite-mediated oxidation of Tyr¹⁰ in A β [14]. Nitration of A β Tyr¹⁰ accelerated its aggregation and plaque formation, and was detected in the core of A β plaques of AD brains [14]. Therefore, the endogenous nitration of A β by A β_{1-16} -heme complex in the presence of H_2O_2 and NaNO_2 , are likely more deleterious than, the exogenous nitration of external proteins (such as glyceraldehyde-3-phosphate dehydrogenase and enolase) by A β -heme/ H_2O_2 / NaNO_2 which have been recently reported [19,20]. Similarly, the A β_{1-40} -heme complex also caused oxidation and nitration of the A β Tyr¹⁰ residue through endogenous peroxidative (H_2O_2) and nitrative ($\text{H}_2\text{O}_2/\text{NO}_2^-$) activities (data not shown).

Furthermore, evidence suggests that tyrosine is a redox-active center in electron transfer between the bulk phase and heme [22]. A recent study indicates that Tyr¹⁰ in A β provides an electron during the Cu^+ or heme(Fe^{2+}) bound A β -mediated $2e^-$ reduction of O_2 to H_2O_2 [6,7]. However, our study showed that the endogenous reactivity of A β -heme peroxidase on the Tyr¹⁰ residue led to the formation of dityrosine cross-linking-contained dimeric peptide, while Tyr¹⁰ was competitively nitrated when nitrite was also present (Fig. 4). Both modifications strongly enhanced the aggregation of A β peptides [5,14,15], thus the formation of A β -heme peroxidase could indeed be a factor contributing to neuroinflammation and A β aggregation, different from the classic inhibitive effect of heme on A β aggregation through A β -heme complex formation [8–11]. These findings shed light on the endogenous peroxidative

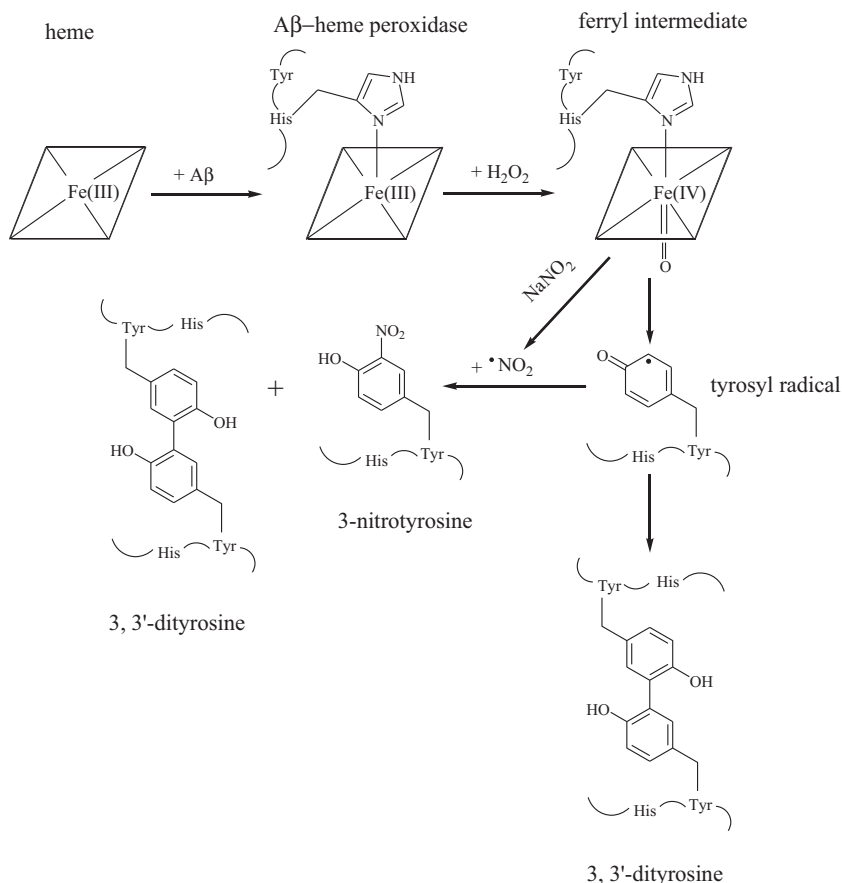


Fig. 4. A β -heme peroxidase induced dimerization and nitration of A β under oxidative stress. Heme bound to A β and formed A β -heme peroxidase complexes. Subsequently, the A β -heme complexes caused endogenous oxidation and nitration of the A β Tyr¹⁰ residue under oxidative (H_2O_2) and nitrative ($\text{H}_2\text{O}_2/\text{NO}_2^-$) stress.

and nitrative activities of A β -heme peroxidase, which would open new directions for better understanding the important roles of A β -heme complex in the pathogenesis of AD.

In summary, we established that His¹³, His¹⁴ and Arg⁵ were essential in peroxidase activity of A β -heme complexes. Under oxidative (H₂O₂) and nitrative (H₂O₂/NO₂⁻) stress conditions, the A β -heme complexes caused oxidation and nitration of the A β Tyr¹⁰ residue through promoting peroxidase-like reactions, where both modifications were strongly associated with the aggregation of A β peptides [5,14,15]. The fact that the three key residues (Arg⁵, Tyr¹⁰ and His¹³) involved in the active site of human A β -heme peroxidase activity are absent in rodent A β peptides, is interesting, as rodent A β -heme complex lacks peroxidase activity and these rodents do not show AD. This further stresses the importance of these residues as well as human A β -heme peroxidase in the neuropathology of AD.

Conflict of interest

The authors declare no competing financial interest.

Acknowledgments

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References

- [1] A. Rauk, *Chem. Soc. Rev.* 38 (2009) 2698–2715.
- [2] D.J. Selkoe, *Phys. Rev.* 81 (2001) 741–766.
- [3] I.W. Hamley, *Chem. Rev.* 112 (2012) 5147–5192.
- [4] X. Zhu, B. Su, X. Wang, M.A. Smith, G. Perry, *Cell Mol. Life Sci.* 64 (2007) 2202–2210.
- [5] C.S. Atwood, G. Perry, H. Zeng, Y. Kato, W.D. Jones, K.Q. Ling, X. Huang, R.D. Moir, D. Wang, L.M. Sayre, M.A. Smith, S.G. Chen, A.I. Bush, *Biochemistry* 43 (2004) 560–568.
- [6] D. Pramanik, C. Ghosh, S.G. Dey, *J. Am. Chem. Soc.* 133 (2011) 15545–15552.
- [7] A.S. Pithadia, M.H. Lim, *Curr. Opin. Chem. Biol.* 16 (2012) 67–73.
- [8] H. Atamna, W.H. Frey II., *Proc. Natl. Acad. Sci. USA* 101 (2004) 11153–11158.
- [9] H. Atamna, K. Boyle, *Proc. Natl. Acad. Sci. USA* 103 (2006) 3381–3386.
- [10] Q. Bao, Y. Luo, W. Li, X. Sun, C. Zhu, P. Li, Z.-X. Huang, X. Tan, *J. Biol. Inorg. Chem.* 16 (2011) 809–816.
- [11] C. Yuan, Z. Gao, *Chem. Res. Toxicol.* 26 (2013) 262–269.
- [12] H. Atamna, W.H. Frey II., N. Ko, *Arch. Biochem. Biophys.* 487 (2009) 59–65.
- [13] M. Paoli, J. Marles-Wright, A. Smith, *DNA Cell Biol.* 21 (2002) 271–280.
- [14] M.P. Kummer, M. Hermes, A. Delekarte, T. Hammerschmidt, S. Kumar, D. Terwel, J. Walter, H.C. Pape, S. König, S. Roeber, F. Jessen, T. Klockgether, M. Korte, M.T. Heneka, *Neuron* 71 (2011) 833–844.
- [15] G. Thiabaud, S. Pizzocaro, R. Garcia-Serres, J.M. Latour, E. Monzani, L. Casella, *Angew. Chem. Int. Ed. Engl.* 52 (2013) 8041–8044.
- [16] D. Pramanik, S.G. Dey, *J. Am. Chem. Soc.* 133 (2011) 81–87.
- [17] Y. Zhou, J. Wang, L. Liu, R. Wang, X. Lai, M. Xu, *ACS Chem. Neurosci.* 4 (2013) 535–539.
- [18] D. Pramanik, G. Ghosh, S. Mukherjee, S.G. Dey, *Coord. Chem. Rev.* 257 (2013) 81–92.
- [19] C. Yuan, L. Yi, Z. Yang, Q. Deng, Y. Huang, H. Li, Z. Gao, *J. Biol. Inorg. Chem.* 17 (2012) 197–207.
- [20] C. Yuan, H. Li, Z. Gao, *J. Biol. Inorg. Chem.* 17 (2012) 1083–1091.
- [21] R. Radi, *Proc. Natl. Acad. Sci. USA* 101 (2004) 4003–4008.
- [22] B.J. Reeder, F. Cutruzzola, M.G. Bigotti, R.C. Hider, M.T. Wilson, *Free Radical Biol. Med.* 44 (2008) 274–283.